

XU GROUP
Department of Chemistry, Peking University

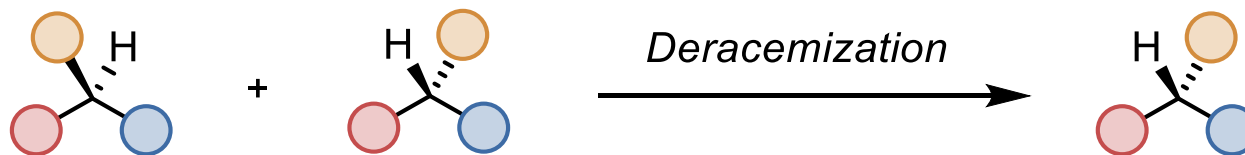
Selected Weekly Literature Presentations

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Deracemization reaction in organic synthesis



Reporter: Ken Chen (陈垦)

Supervisor: Yan Xu (许言)

2023.03.17

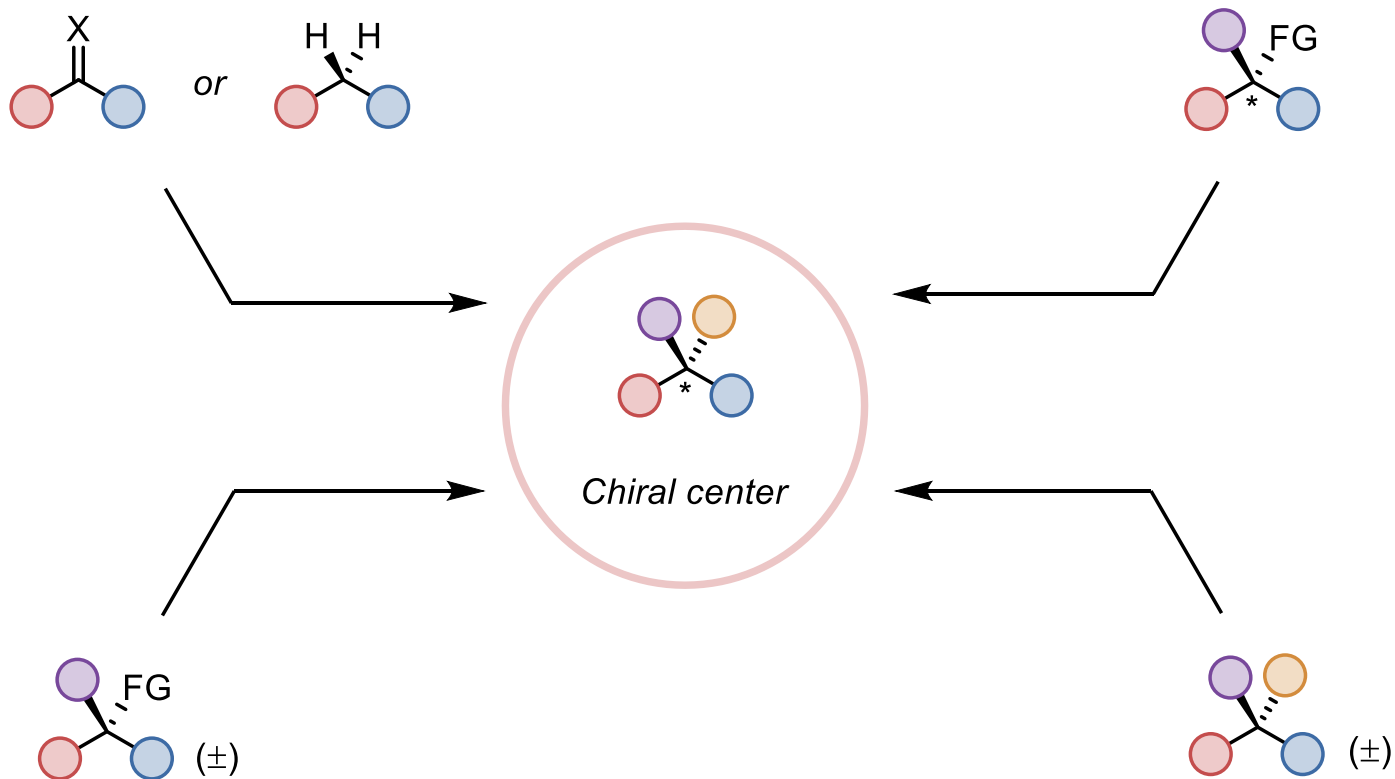
Content

- Introduction of cyclic deracemization
- Deracemization by chemical methods
- Enzyme catalyzed deracemization
- Light-empowered deracemization

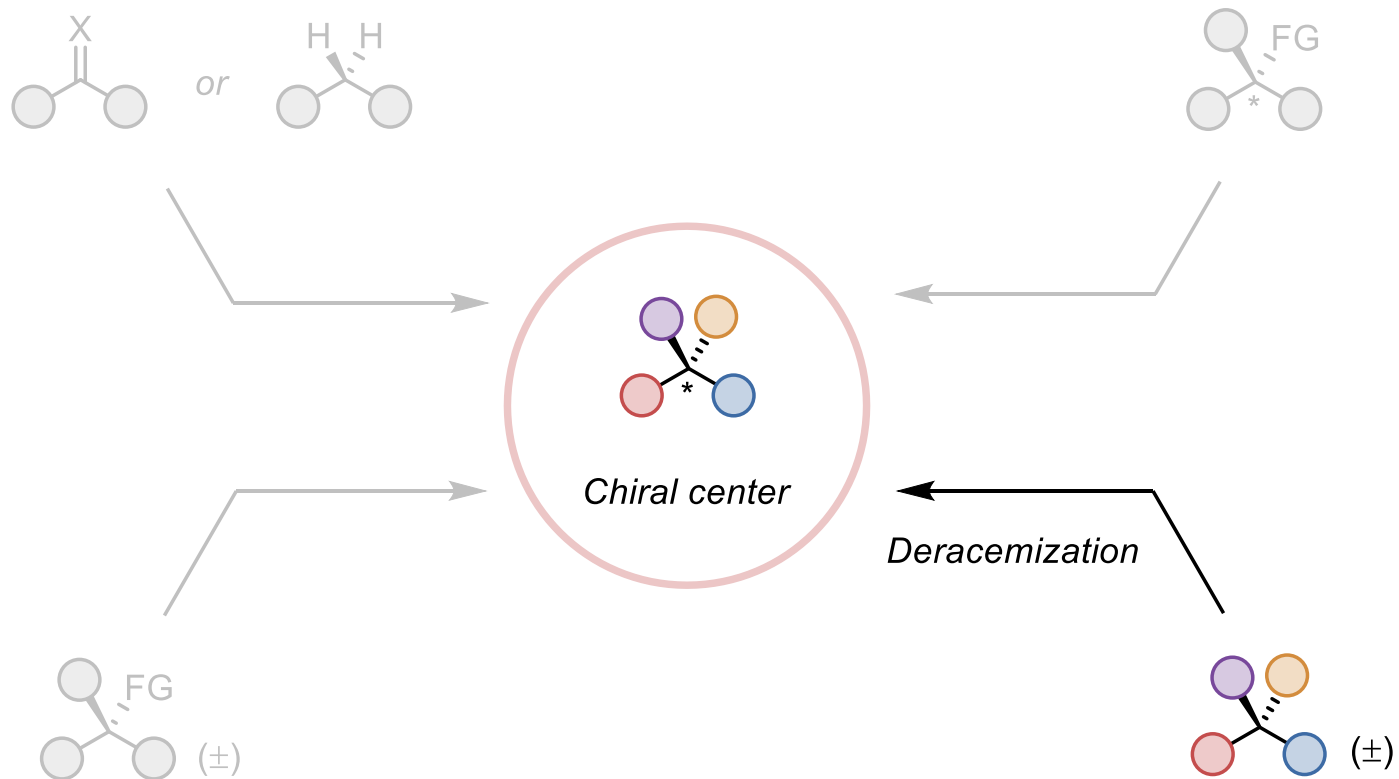
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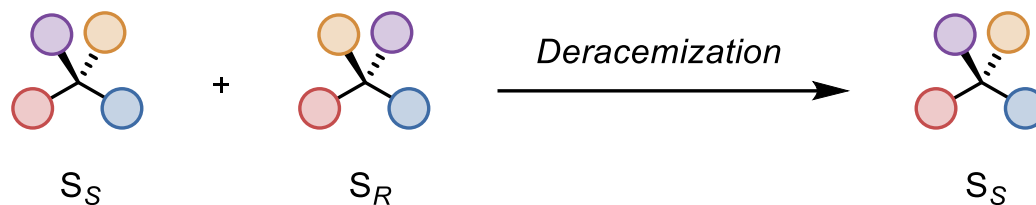
Forming a chiral center



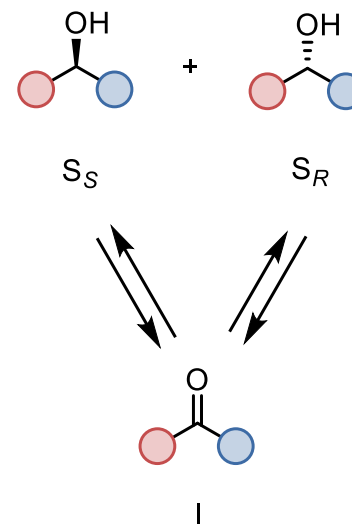
Deracemization



Deracemization modes



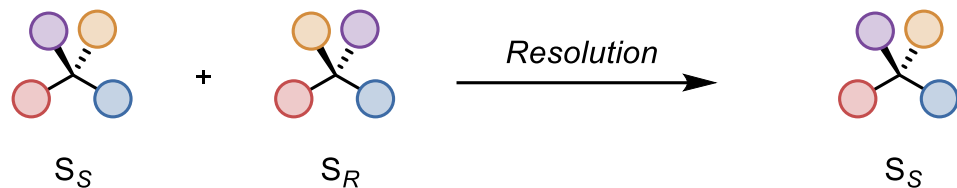
For example:



- Find an intermediate
- Construct two converse processes

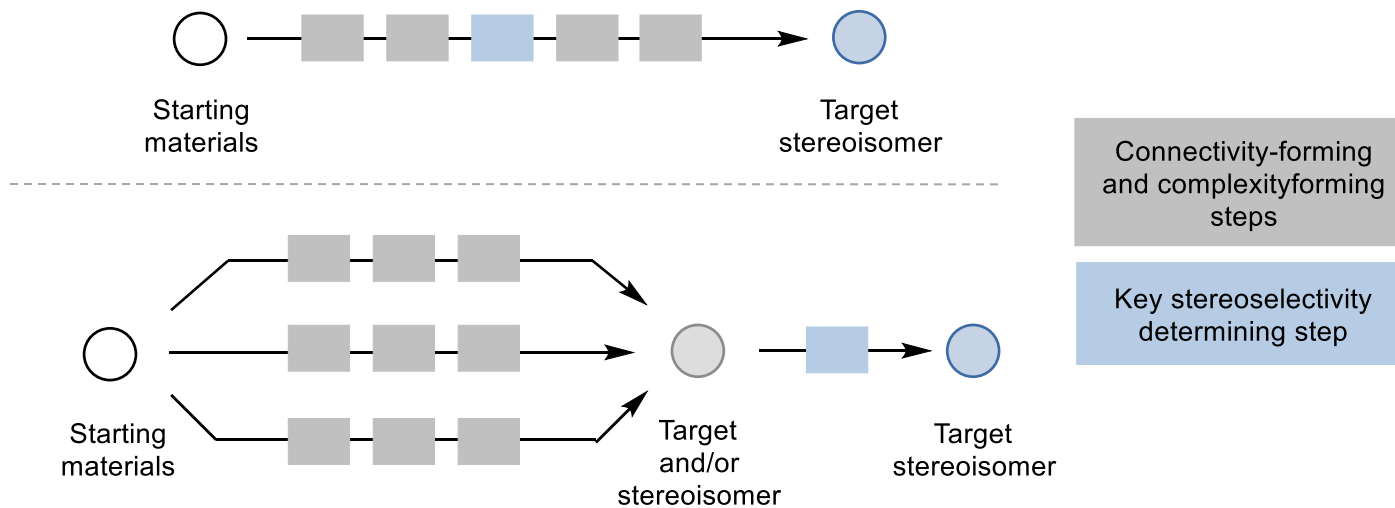
Advantages of deracemization

■ 100% theoretical yield



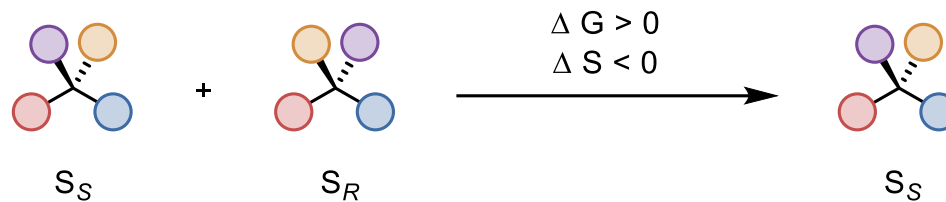
50% theoretical yield

■ Stereochemical editing at late stage



Challenges of deracemization

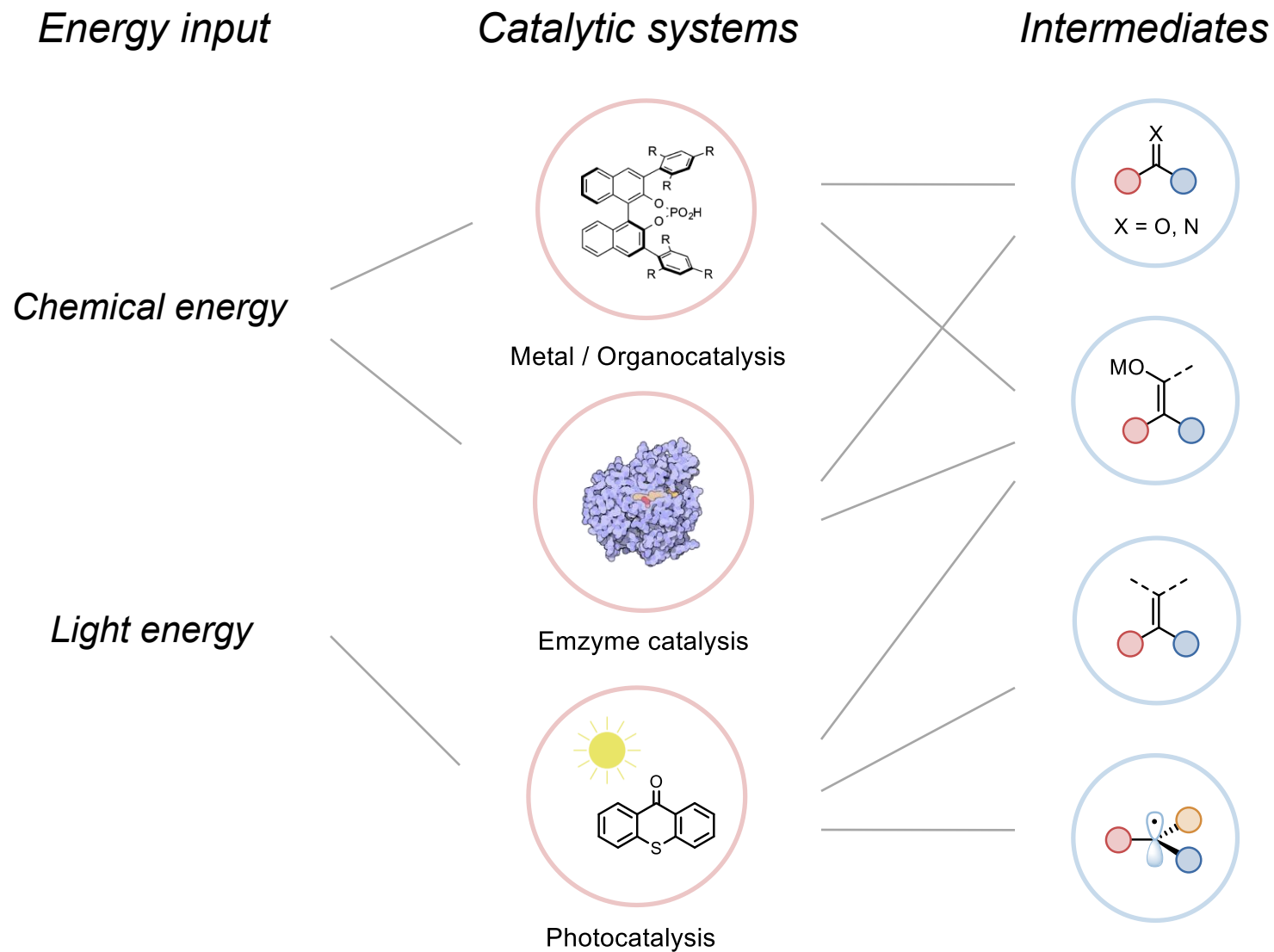
- *Require energy input*



- *'Two processes in one pot'*



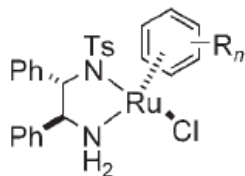
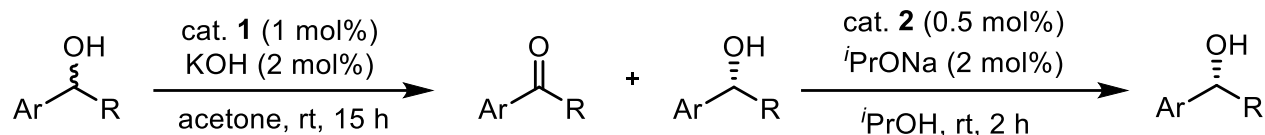
Deracemization



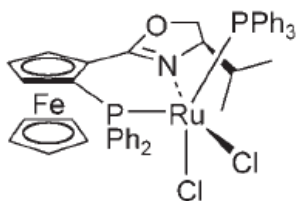
Content

- Introduction of cyclic deracemization
- **Deracemization by chemical methods**
- Enzyme catalyzed deracemization
- Light-empowered deracemization

Deracemization of alcohol in two steps



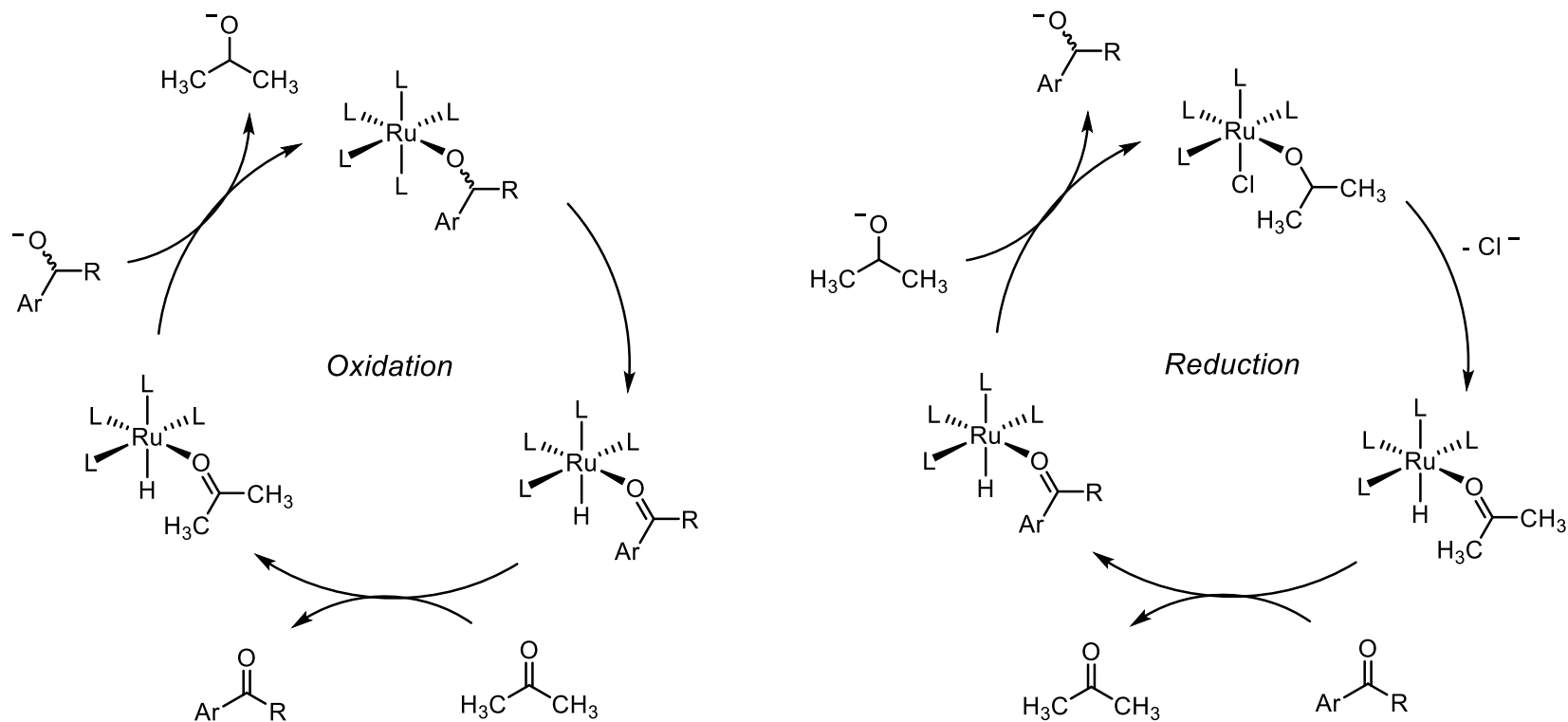
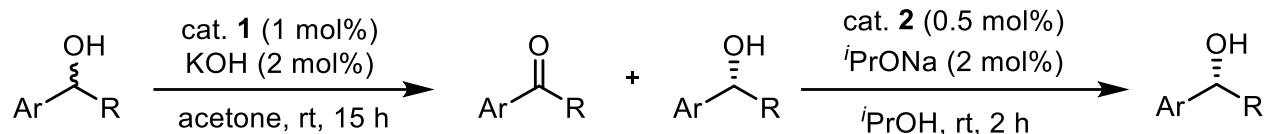
cat. 1, [RuCl{(S,S)-tsdpen}(η⁶-mesitylene)]



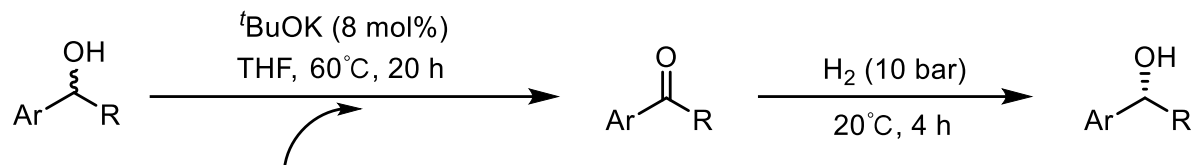
cat. 2, [RuCl₂(PPh₃)(ip-FOXAP)]

| Entry | Ar | R | Recovery of 3 [%] ^[b] | ee of 3 [%] ^[c,d] |
|-------------------|-----------------------------------|-------------|---|-------------------------------------|
| 1 | Ph | Me | 90 (3 a) | 94 (<i>R</i>) |
| 2 | 4-MeC ₆ H ₄ | Me | 92 (3 b) | 92 (<i>R</i>) |
| 3 | 4-FC ₆ H ₄ | Me | 82 (3 c) | 90 (<i>R</i>) |
| 4 | 4-ClC ₆ H ₄ | Me | 99 (3 d) | 89 (<i>R</i>) |
| 5 | 4-BrC ₆ H ₄ | Me | 99 (3 e) | 92 (<i>R</i>) |
| 6 ^[e] | 4-PhC ₆ H ₄ | Me | 99 (3 f) | 93 (<i>R</i>) |
| 7 ^[f] | 3-MeC ₆ H ₄ | Me | 81 (3 g) | 91 (<i>R</i>) |
| 8 | 2-MeC ₆ H ₄ | Me | 93 (3 h) | 2 (<i>R</i>) |
| 9 ^[e] | Ph | Et | 88 (3 i) | 95 (<i>R</i>) |
| 10 | Ph | Et | > 95 (3 i) | 88 (<i>R</i>) |
| 11 ^[e] | Ph | <i>n</i> Pr | 85 (3 j) | 90 (<i>R</i>) |
| 12 | Ph | <i>n</i> Pr | 96 (3 j) | 81 (<i>R</i>) |
| 13 ^[e] | Ph | <i>n</i> Bu | 86 (3 k) | 94 (<i>R</i>) |
| 14 | Ph | <i>n</i> Bu | > 95 (3 k) | 85 (<i>R</i>) |

Deracemization of alcohol in two steps

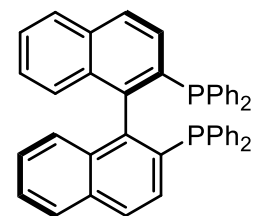


Deracemization of alcohol in two steps

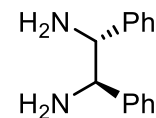


[RuCl₂(benzene)]₂ (1 mol%)
 (R)-BINAP (2 mol%)
 (R,R)-DPEN (2 mol%)
 cyclohexone (2.4 equiv.)
 110 °C, 1 h

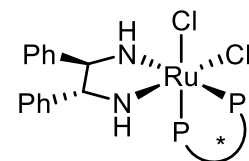
| Alcohol | Ar | R | Yield ^a (%) | ee (%) |
|----------------|--|--------------------------------|------------------------|--------|
| 4 | C ₆ H ₅ | C ₂ H ₅ | 87 | 83 |
| 5 ^b | C ₆ H ₅ | C ₃ H ₇ | 82 | 87 |
| 6 | C ₆ H ₅ | C ₄ H ₉ | 92 | 86 |
| 7 | 2-Naphthyl | C ₅ H ₁₁ | 95 | 82 |
| 8 | <i>m</i> -MeC ₆ H ₄ | C ₅ H ₁₁ | 97 | 90 |
| 9 | <i>p</i> -MeOC ₆ H ₄ | CH ₃ | 89 | 79 |
| 10 | <i>p</i> -MeOC ₆ H ₄ | C ₅ H ₁₁ | 94 | 88 |
| 11 | <i>m</i> -ClC ₆ H ₄ | C ₅ H ₁₁ | 26 | 57 |
| 12 | <i>p</i> -Me ₂ NC ₆ H ₄ | C ₅ H ₁₁ | 96 | 92 |



(R)-BINAP

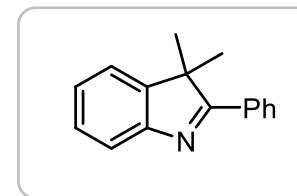
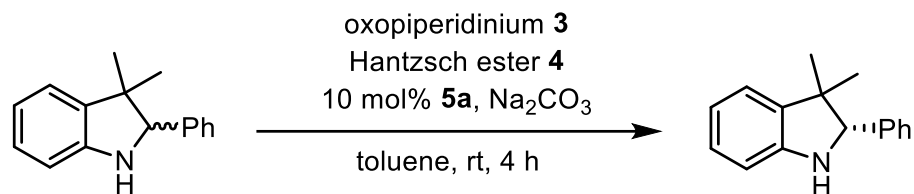


(R,R)-DPEN

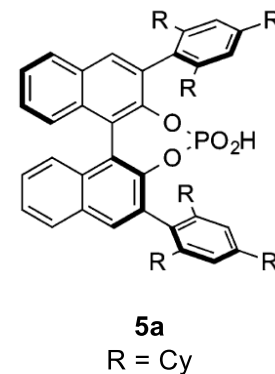
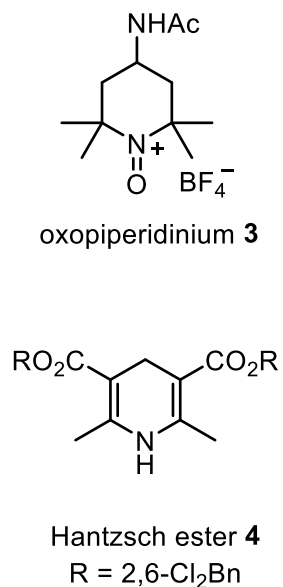
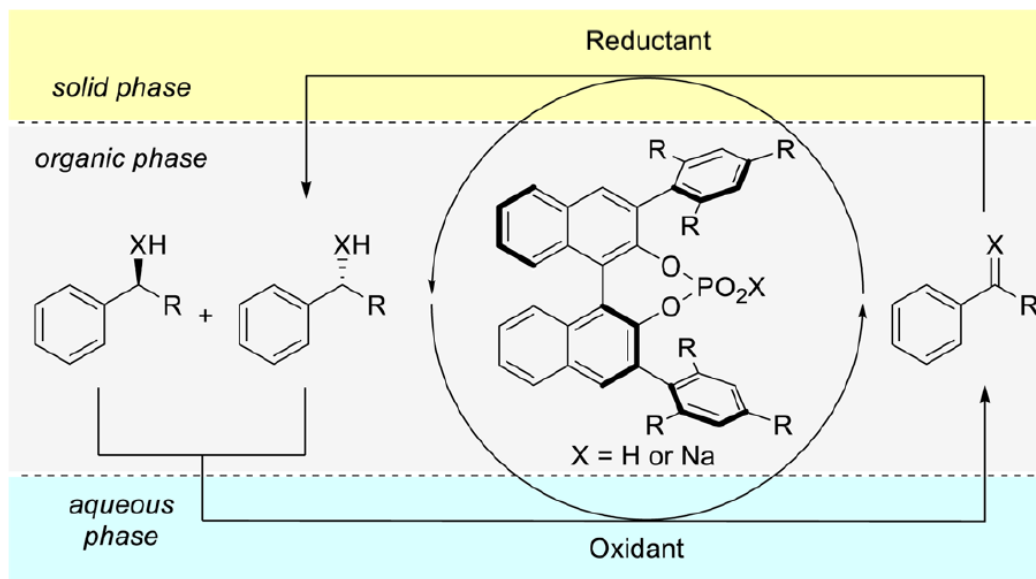


RuCl₂[(R)-BINAP][(R,R)-DPEN]

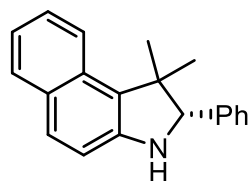
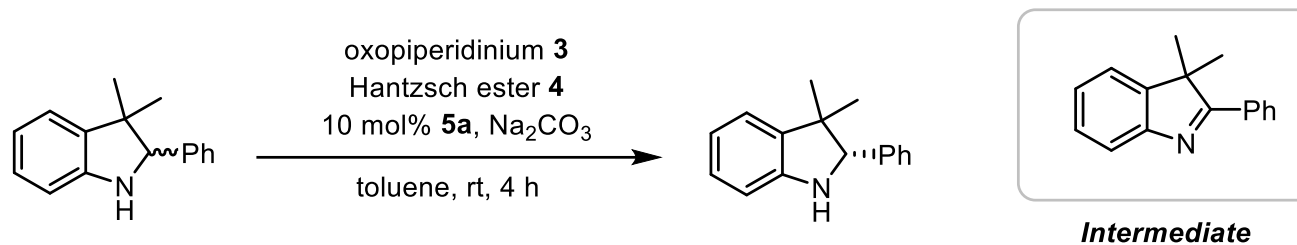
Deracemization of 3H-Indolines in one pot



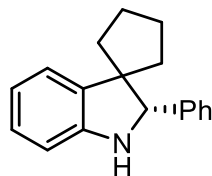
Intermediate



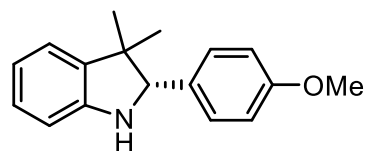
Deracemization of 3H-Indolines in one pot



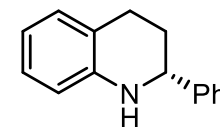
92%, 93% ee



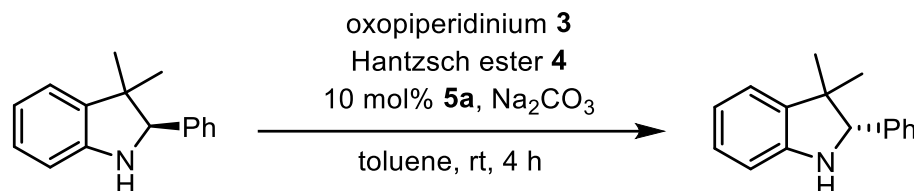
89%, 93% ee



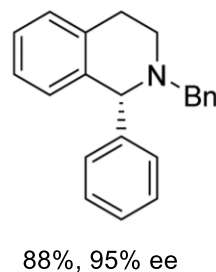
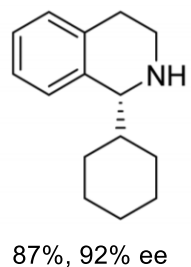
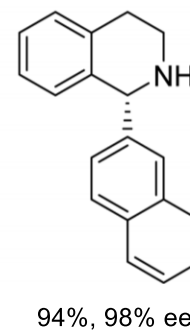
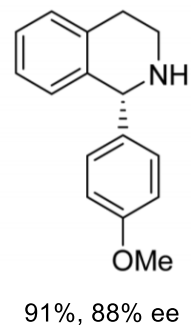
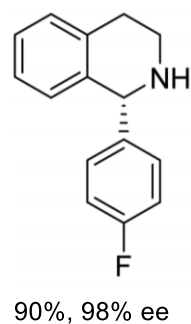
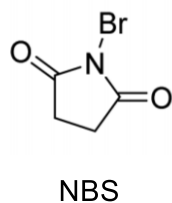
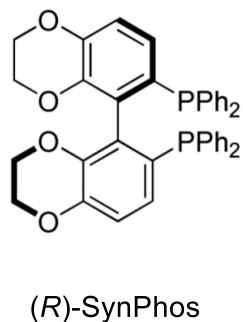
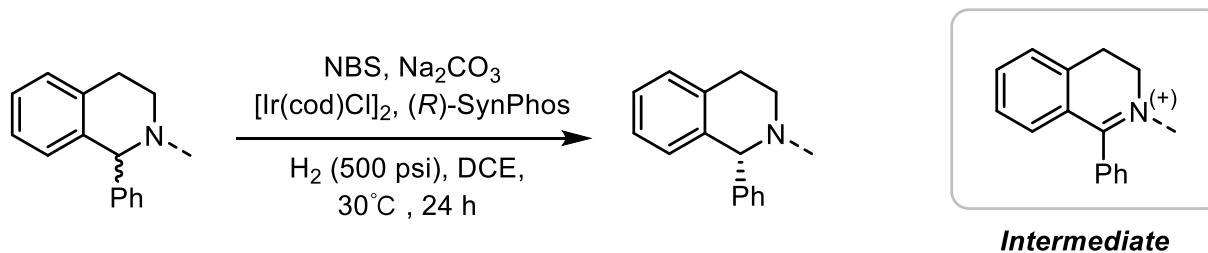
82%, 88% ee



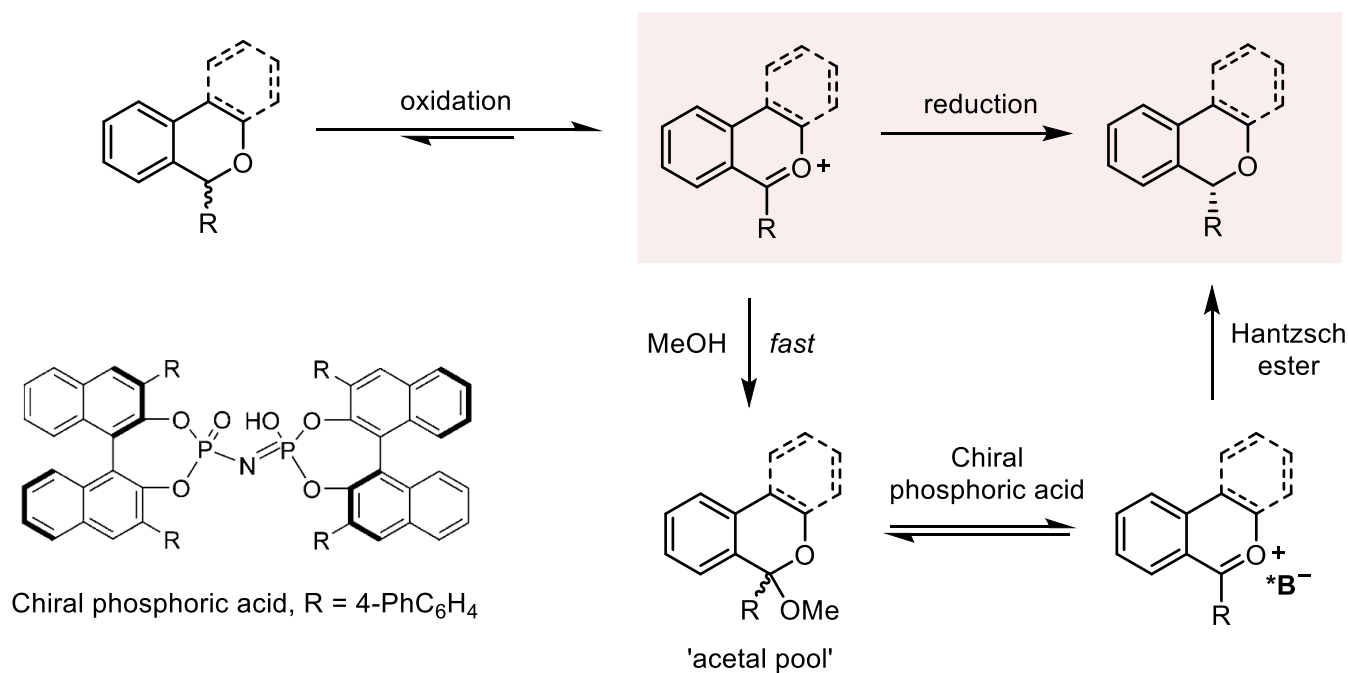
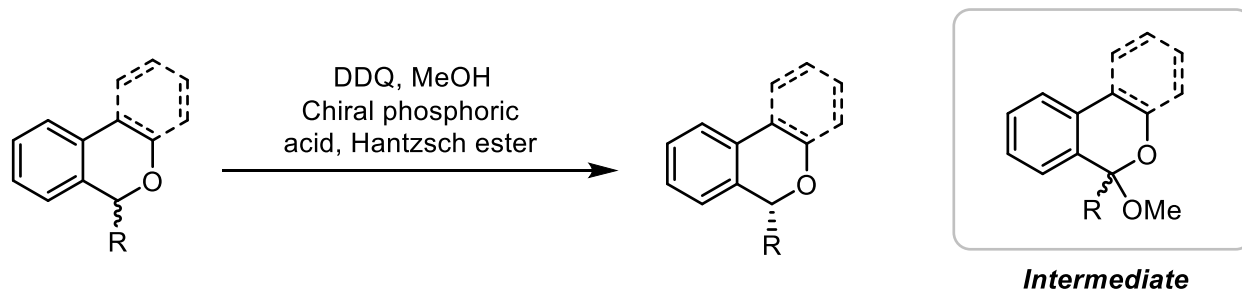
95%, 92% ee



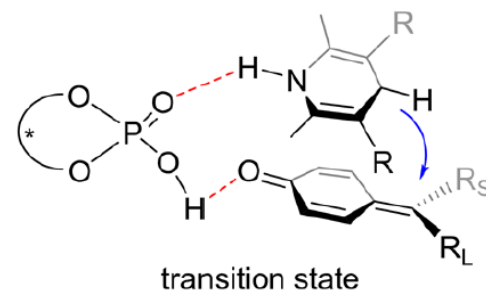
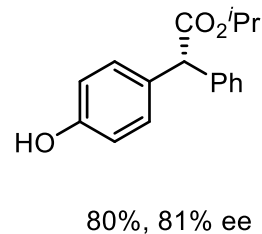
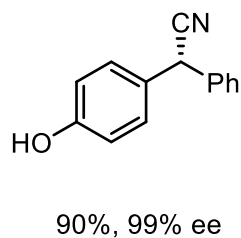
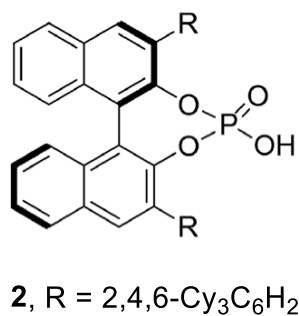
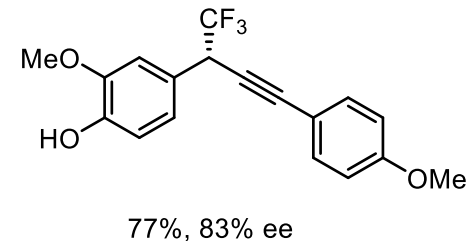
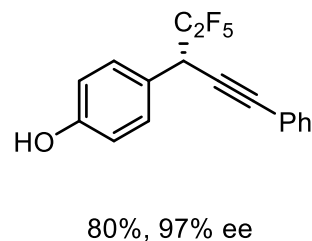
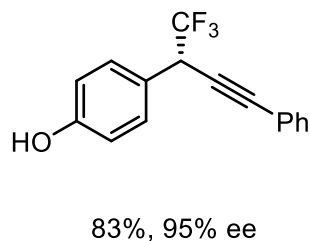
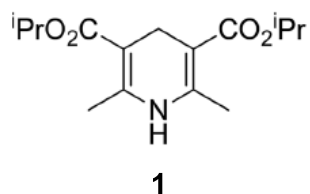
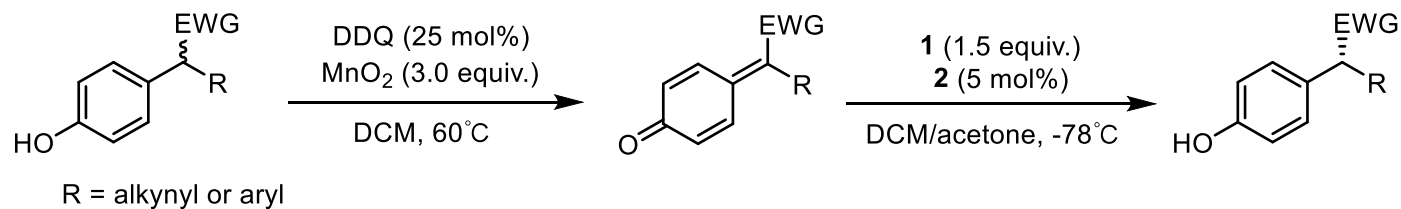
Deracemization of tetrahydroisoquinoline



Deracemization of cyclic benzylic ethers



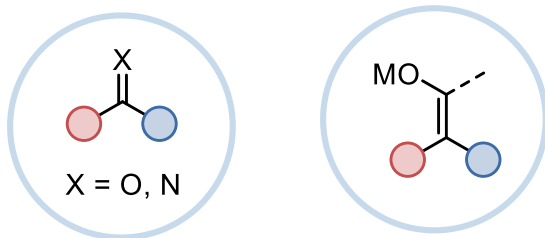
Deracemization via semiquinone intermediate



Summary

■ Chemical method

■ Common intermediate



■ Common process

- Redox
- Proton transfer

■ Energy input

- Chemical energy

■ Stereocontrol

- Chiral ligands
- CPA

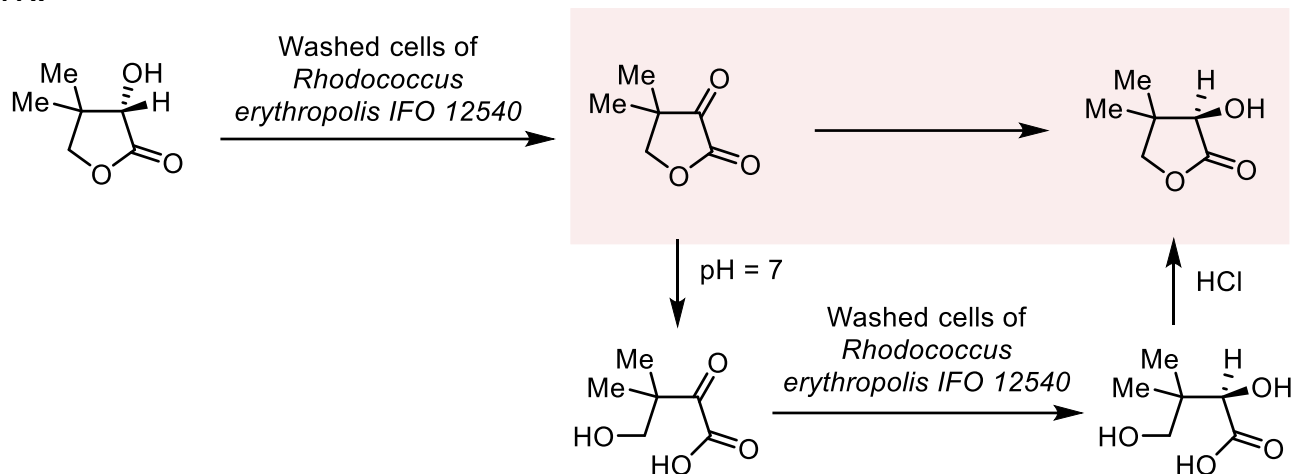
■ Two converse processes
need to be compatible in one
pot

Content

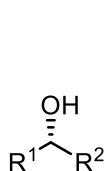
- Introduction of cyclic deracemization
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- Light-empowered deracemization

Deracemization by washed cell

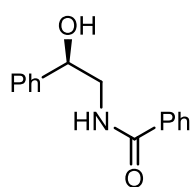
■ First work:



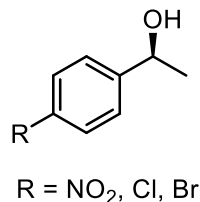
■ Examples (using washed cells):



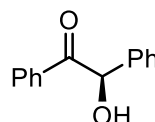
Spingomonas paucimobilis



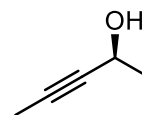
Cunninghamella echinulata



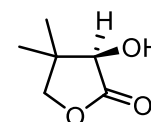
Aspergillus terreus CCT 3320 and CCT 4083



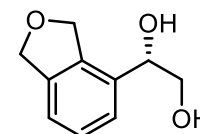
Rhizopus oryzae



Nocardia pseudosporangifera



Rhodococcus erythropolis

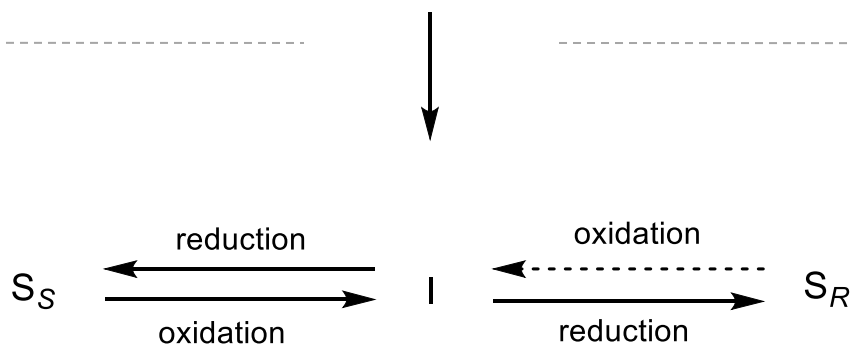
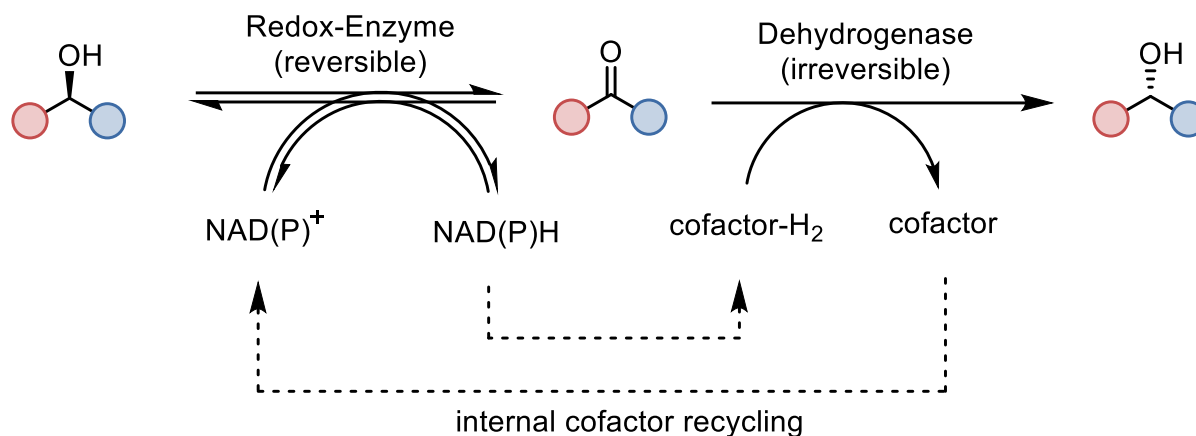


andida boidinii *Pichia methanolica* *Hansenula polymorpha*

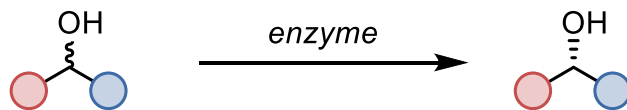
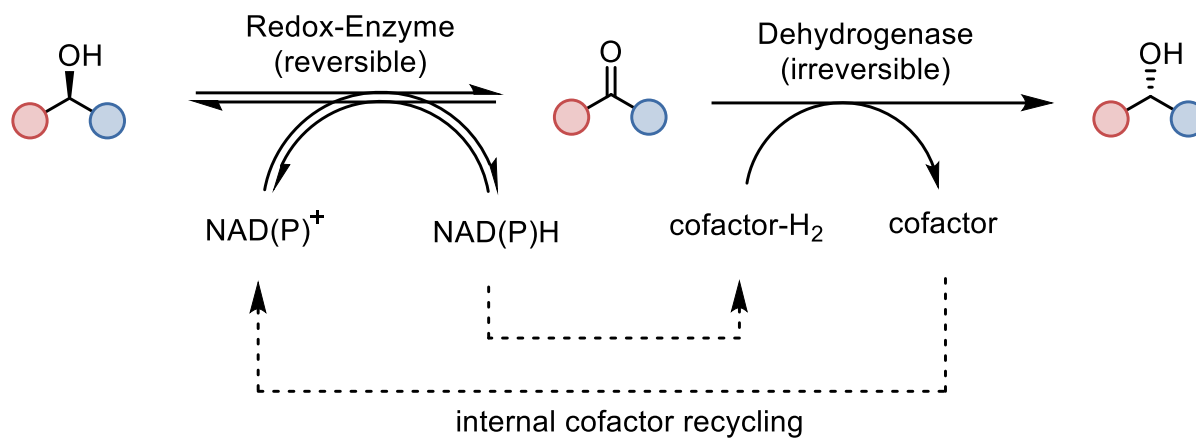
Yamada, H. et al. *Appl. Environ. Microbiol.* **1987**, 53, 519–522.

Kroutil, W et al. *Adv. Synth. Catal.* **2006**, 348, 1789–1805.

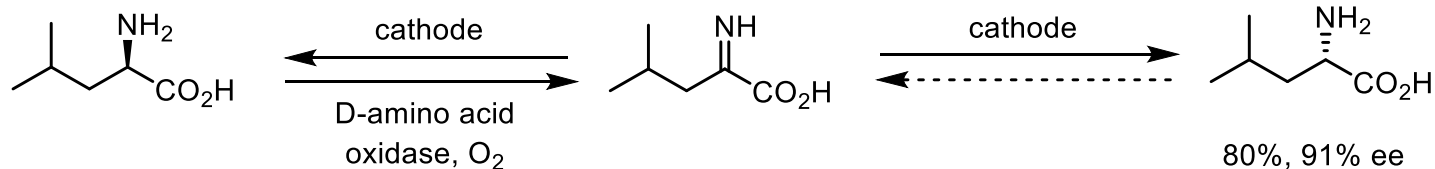
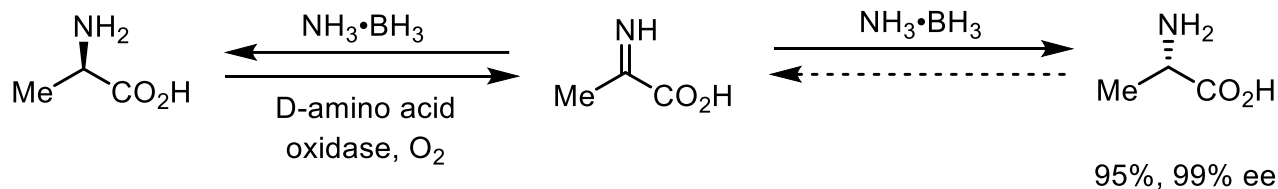
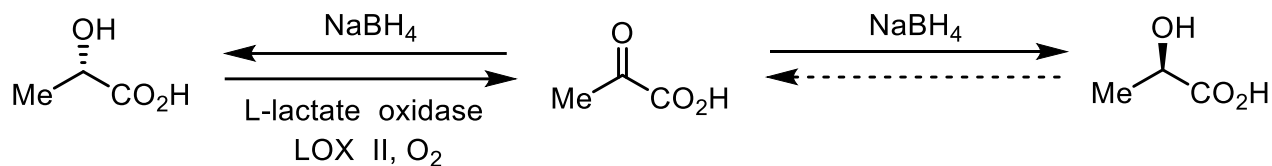
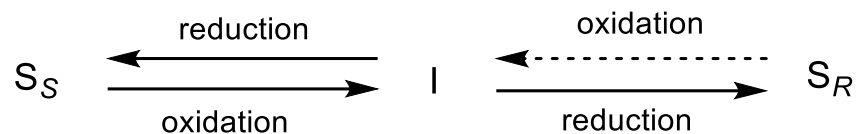
Catalytic mechanism of washed cell



Catalytic mechanism of washed cell

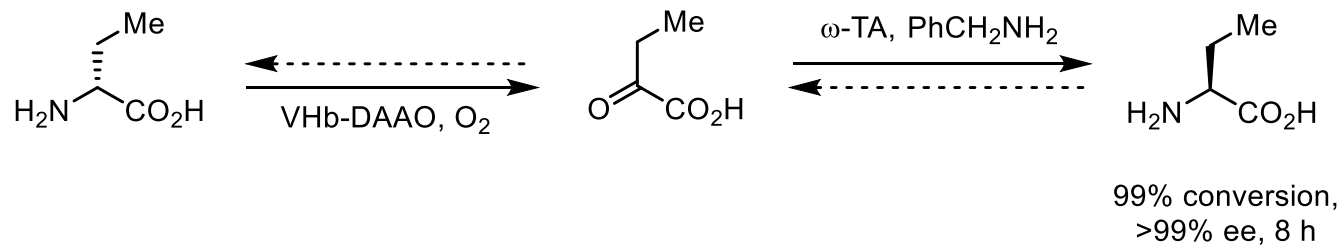
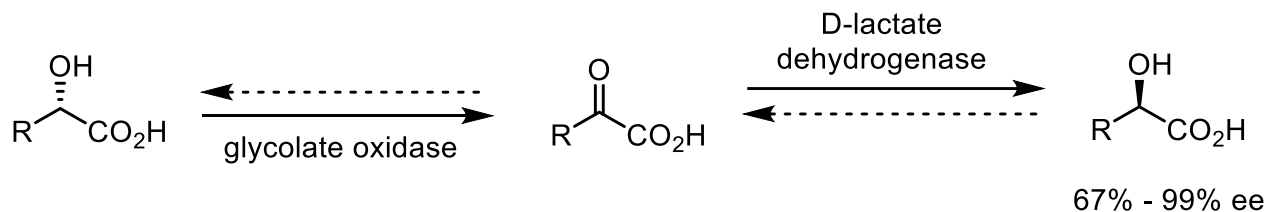
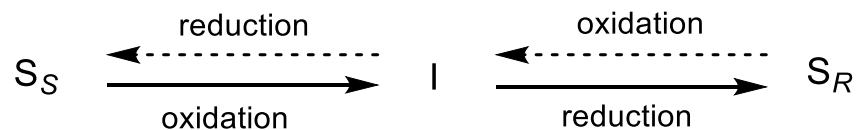


Deracemization by enzyme



Soda, K. *et al. Biotechnol. Bioeng.* **2001**, 73, 80–82.
Turner, N. J. *et al. Biochem. Soc. Trans.* **2006**, 34, 287–290.
Lütz, S. *et al. Electrochim. Acta.* **2008**, 53, 3175–3180.

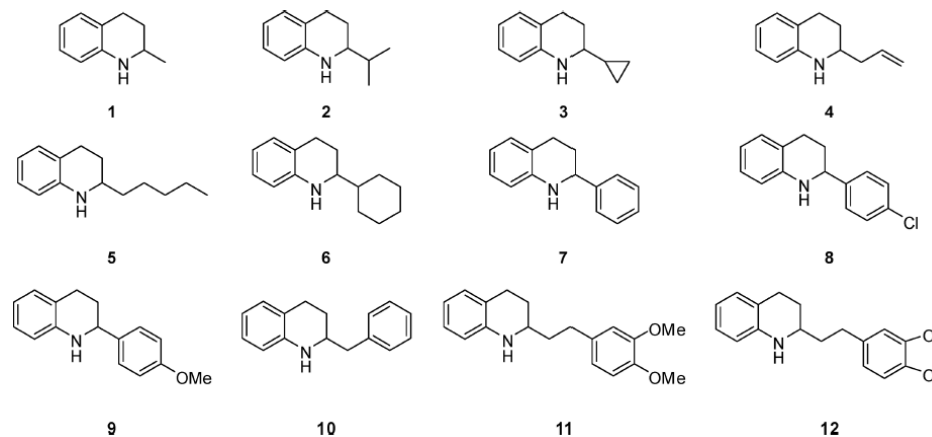
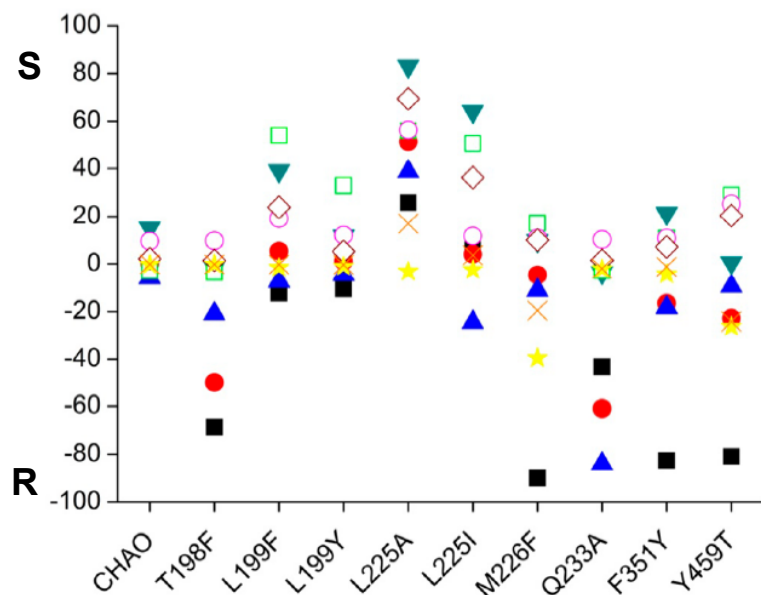
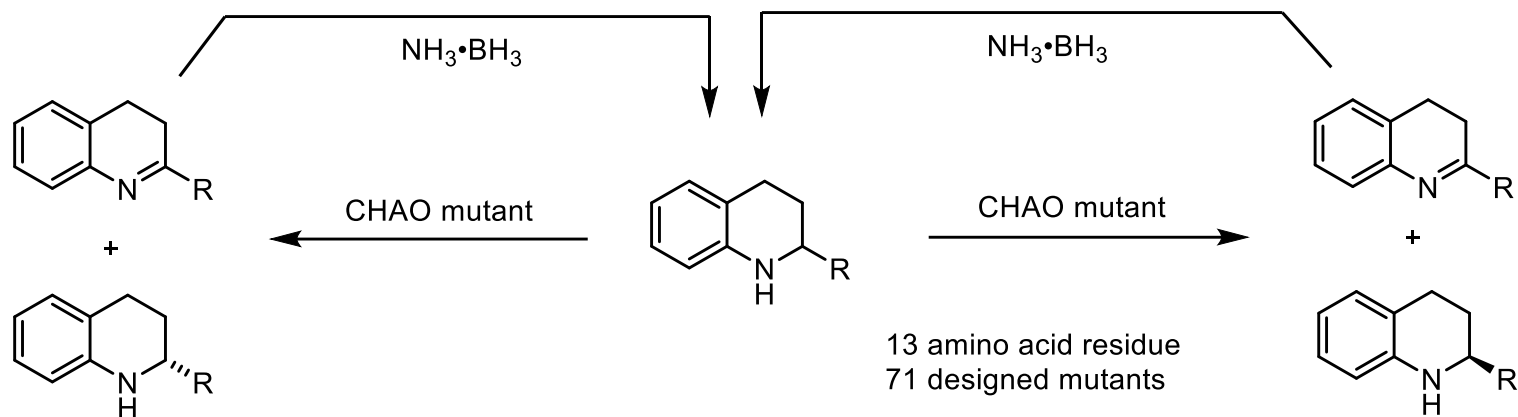
Deracemization by enzyme



Schreier, P. *et al. Tetrahedron: Asymmetry* **1998**, 9, 351–355.

Patel, R. N. *et al. Org. Process Res. Dev.* **2011**, 2, 241.

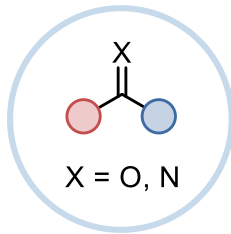
Optimized by mutants



Summary

■ Enzyme catalysis

■ Common intermediate



■ Common process

■ Redox

■ Energy input

■ Chemical energy

■ Washed cell

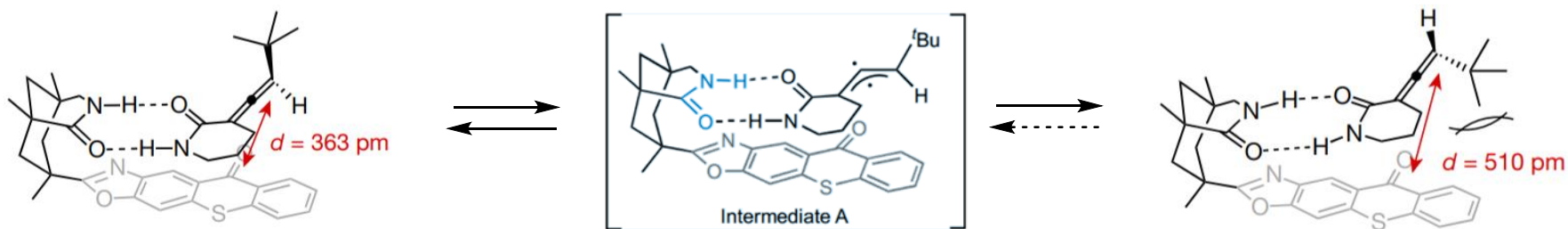
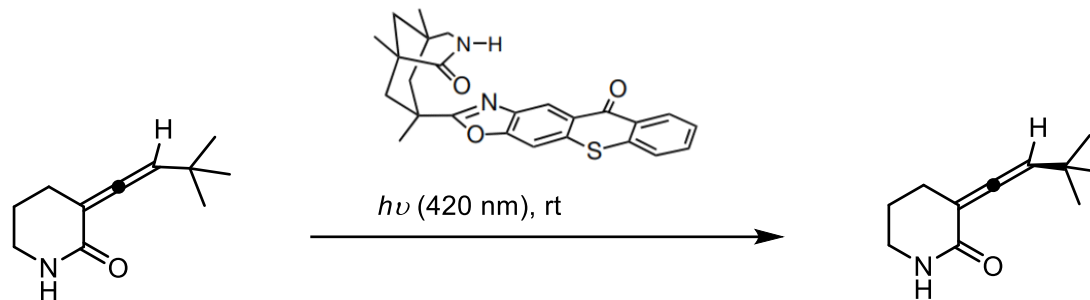
■ Natural enzyme

■ Enzyme mutants

Content

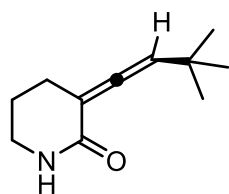
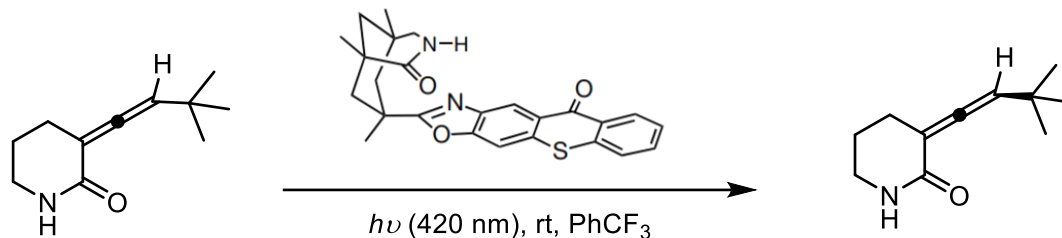
- Introduction of cyclic deracemization
- Deracemization by chemical methods
- Enzyme catalyzed deracemization
- **Light-empowered deracemization**

Deracemization of chiral allenes

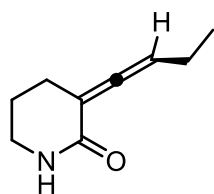


- Low association constants
- Low sensitization efficiencies

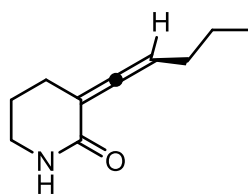
Deracemization of chiral allenes



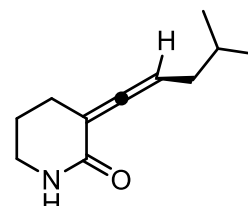
89%, 96% ee



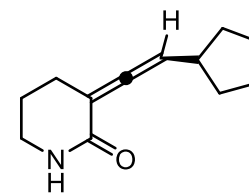
100%, 93% ee



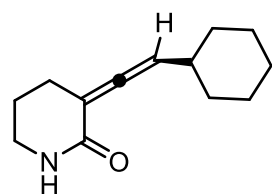
84%, 92% ee



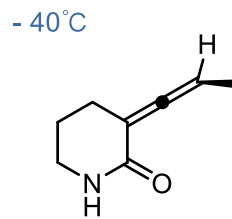
98%, 91% ee



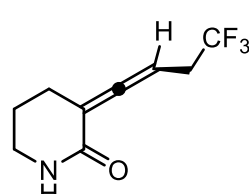
100%, 93% ee



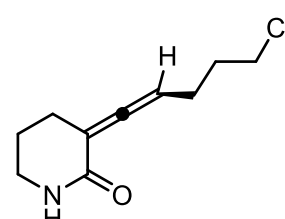
76%, 94% ee



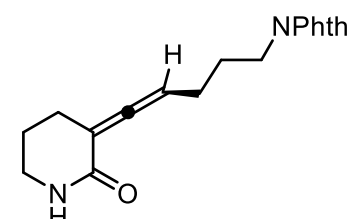
56%, 90% ee



92%, 86% ee

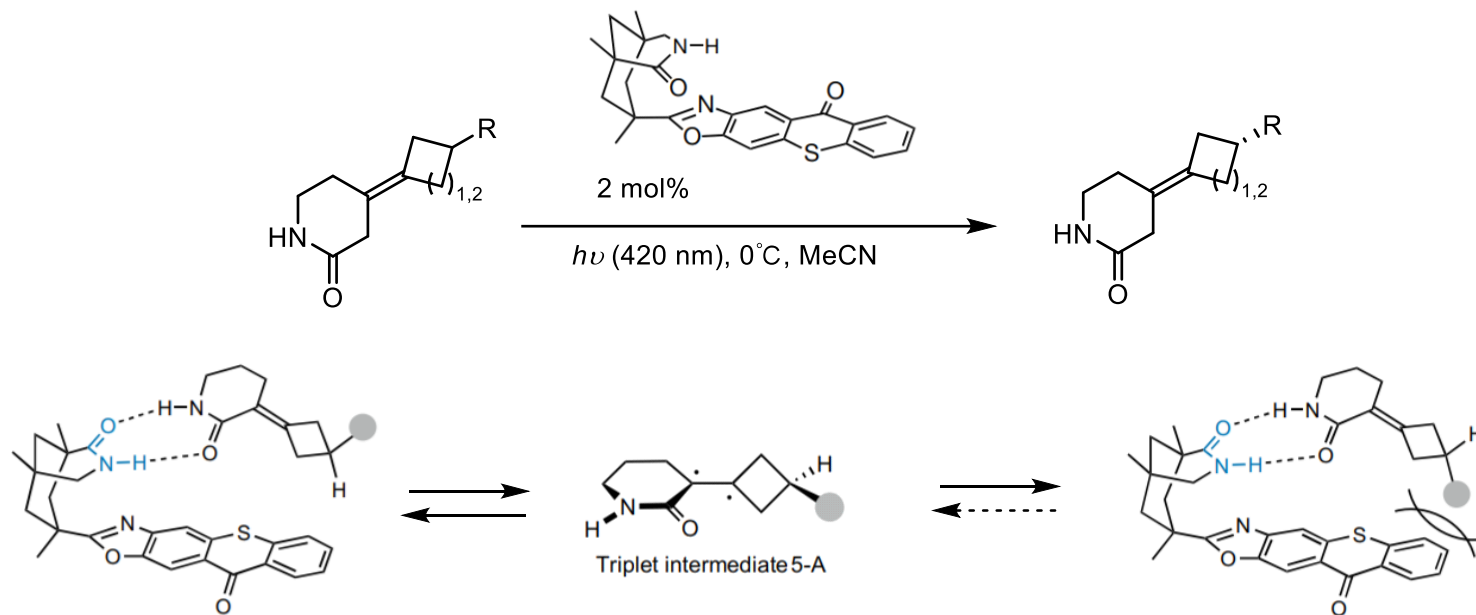
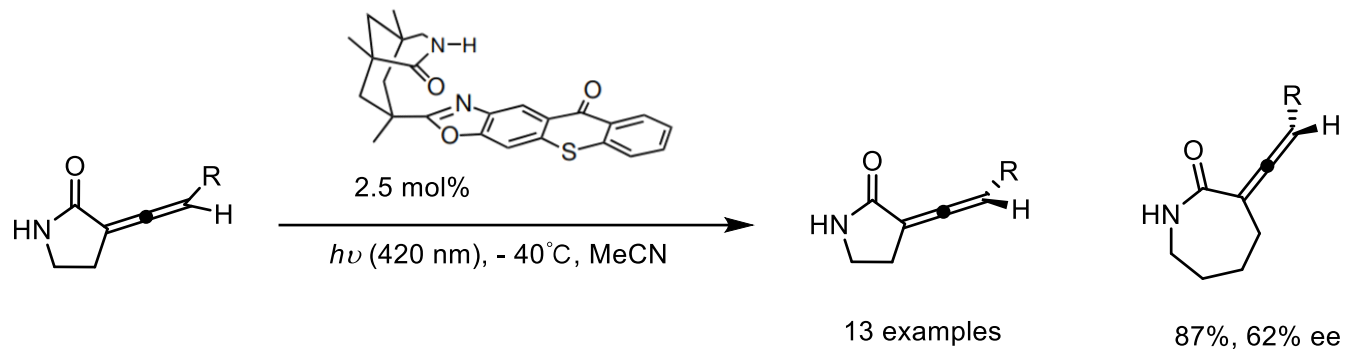


87%, 92% ee



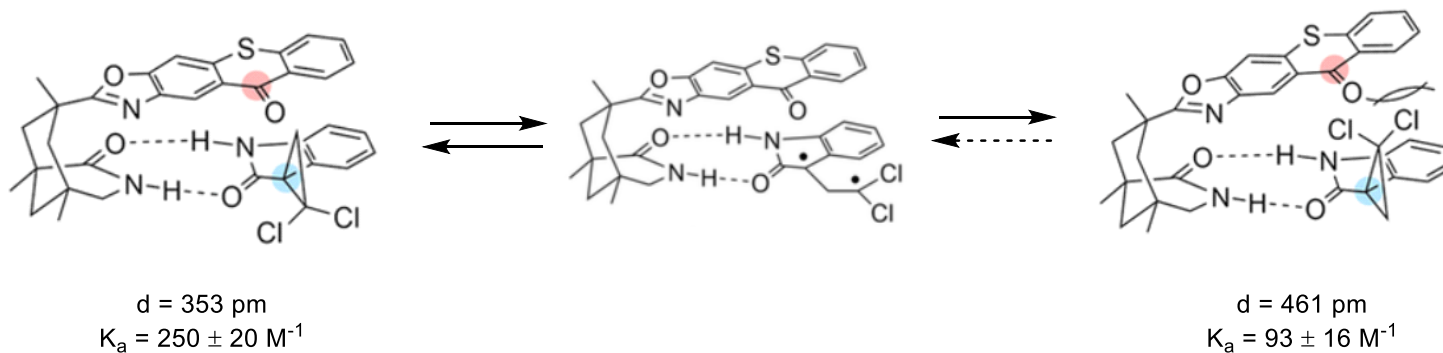
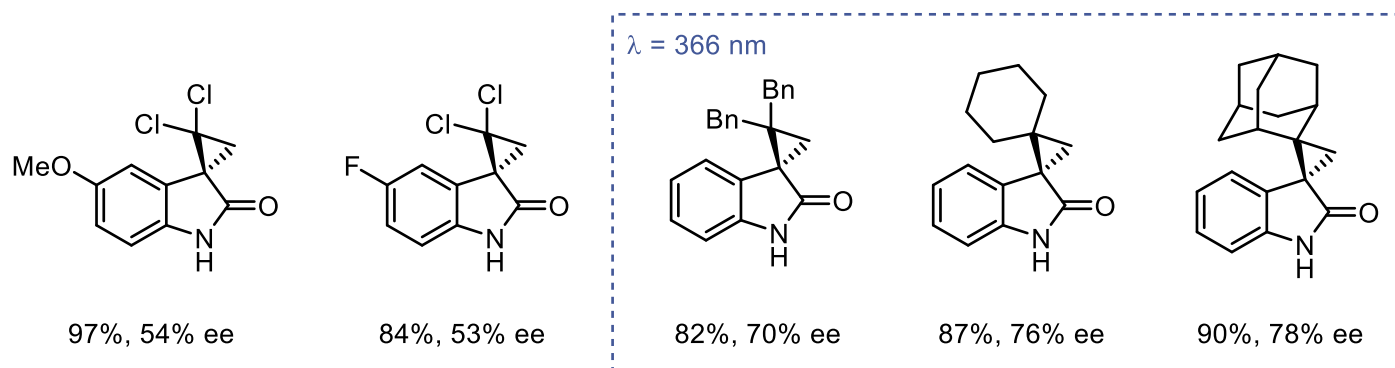
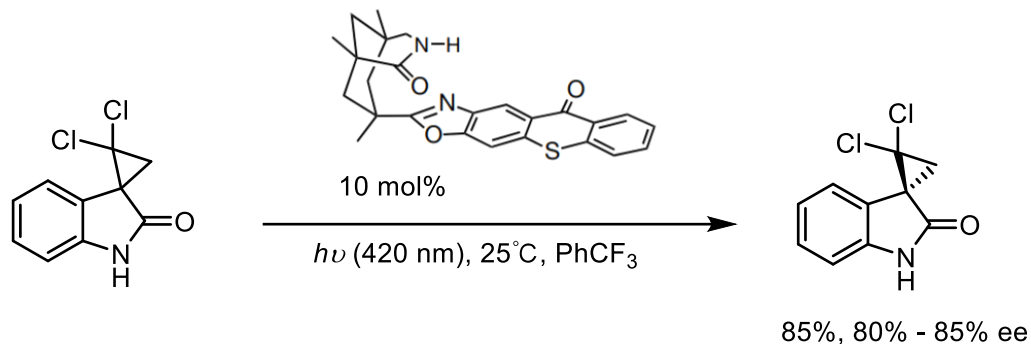
72%, 90% ee

Deracemization of Chiral Alkenes

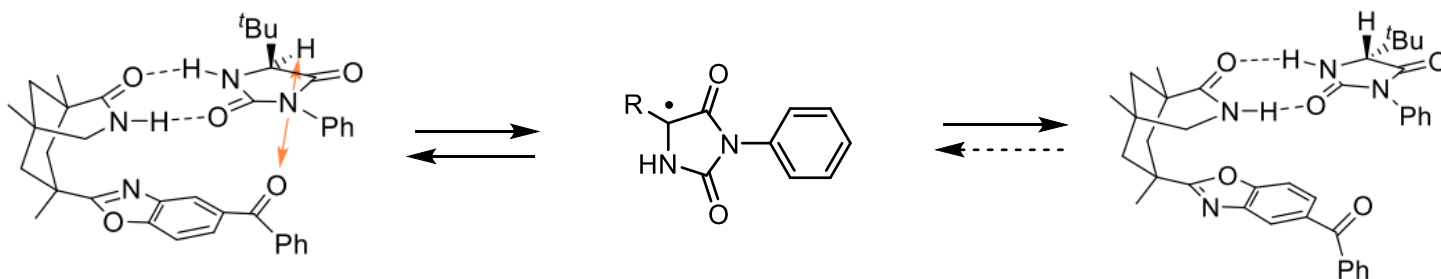
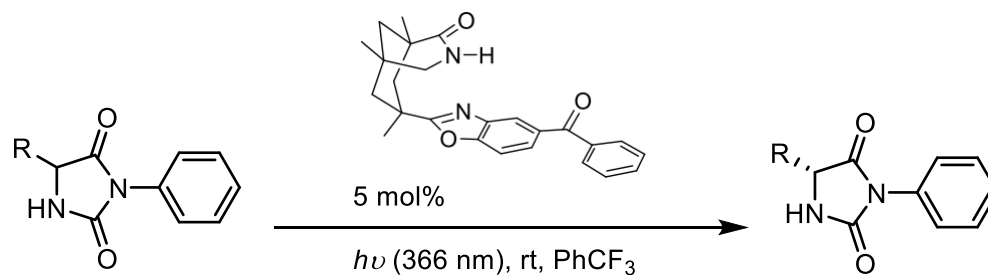
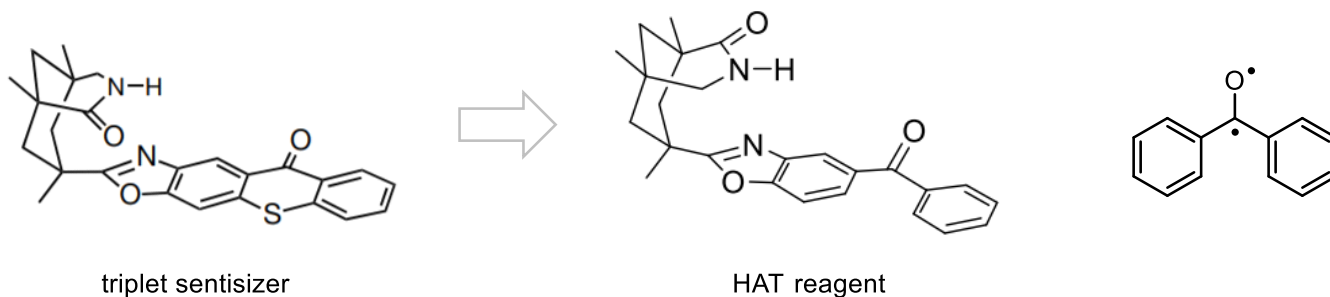


Bach, T. *et al. Angew. Chem. Int. Ed.* **2020**, *59*, 12785–12788.
Bach, T. *et al. J. Am. Chem. Soc.* **2022**, *144*, 10133–10138.

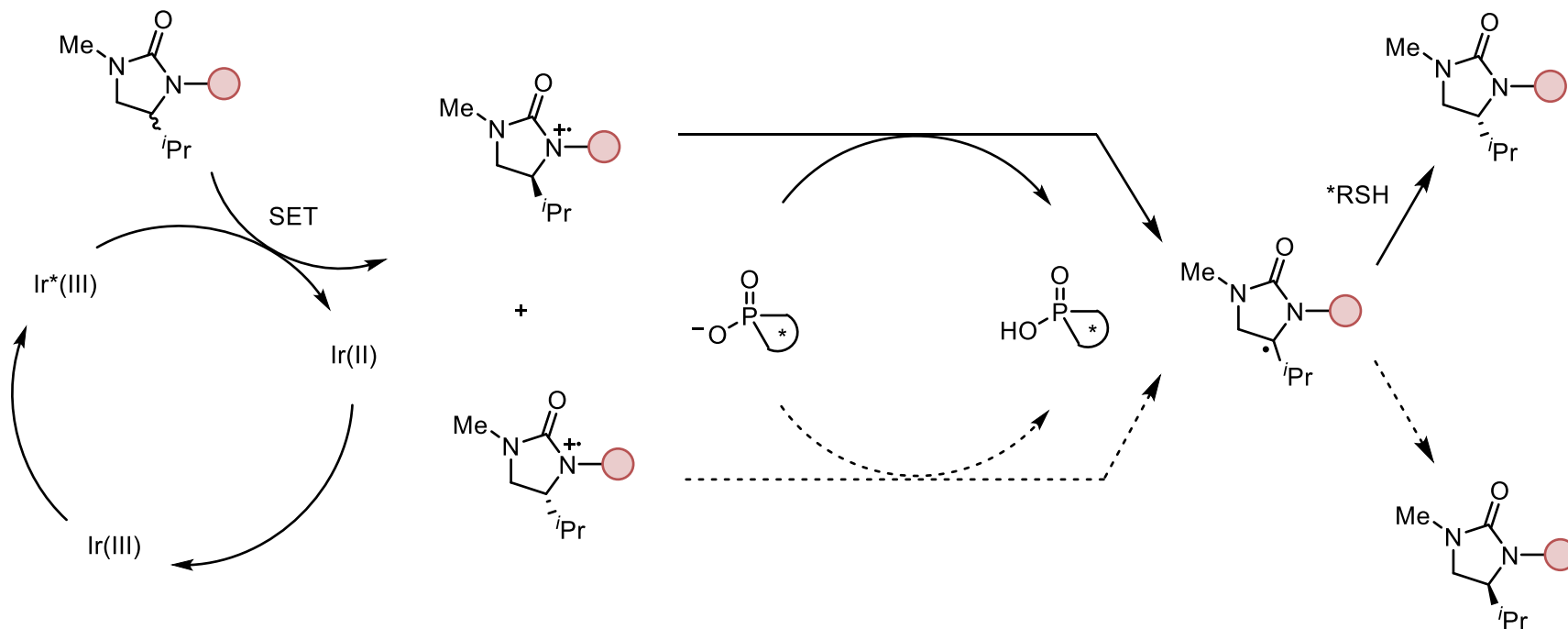
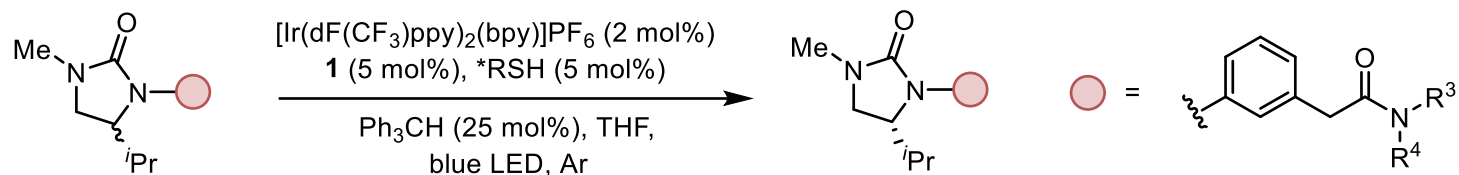
Deracemization of Spirocyclopropyl



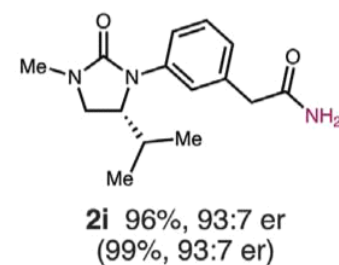
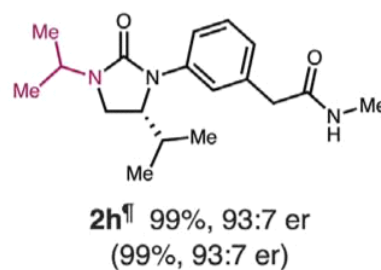
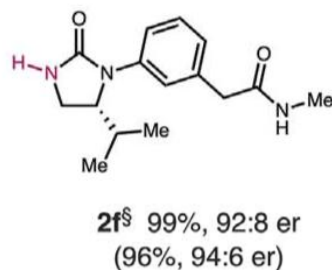
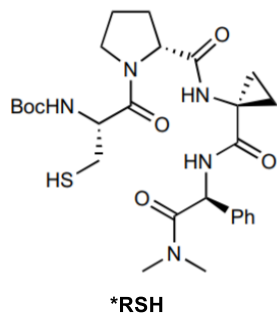
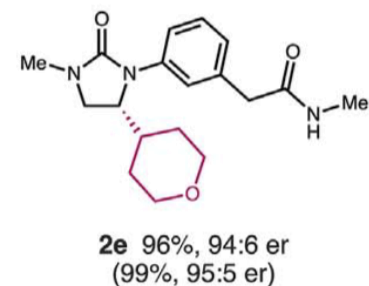
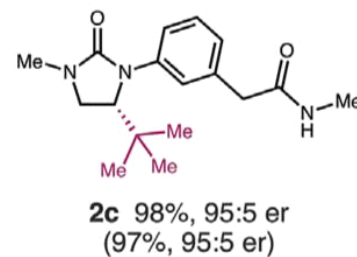
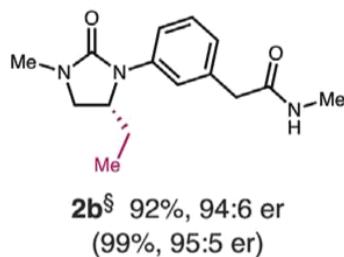
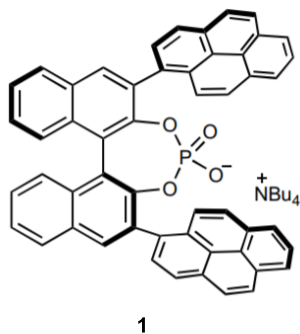
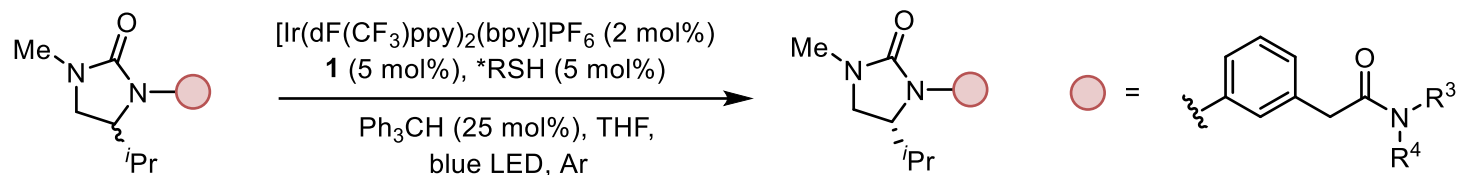
Deracemization at sp^3 -Hybridized Carbon



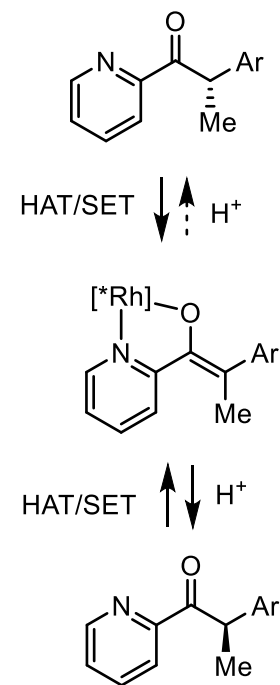
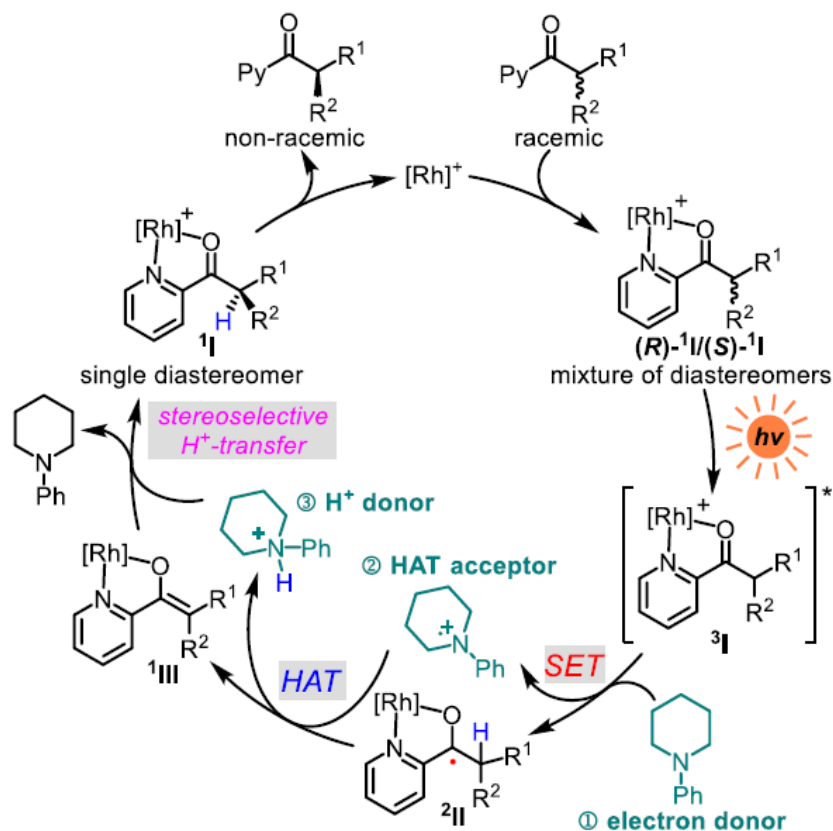
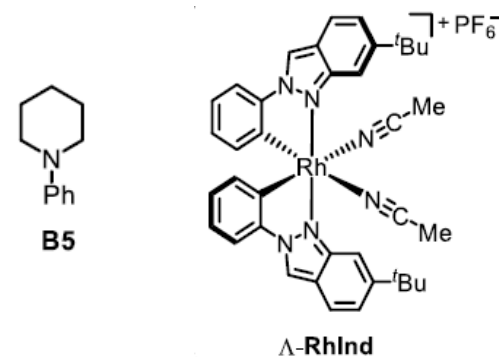
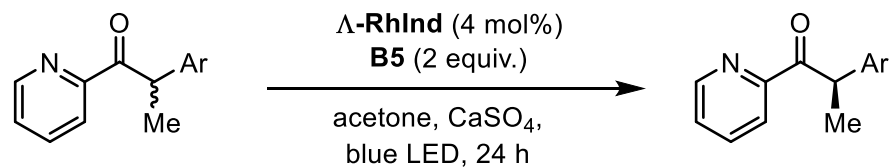
Deracemization at sp^3 -Hybridized Carbon



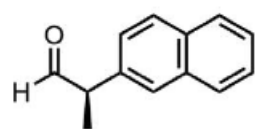
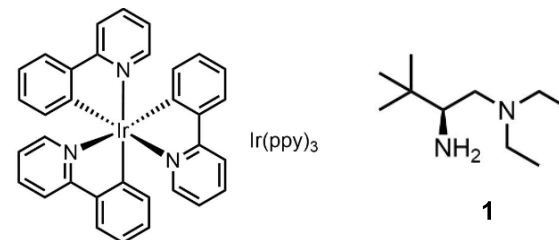
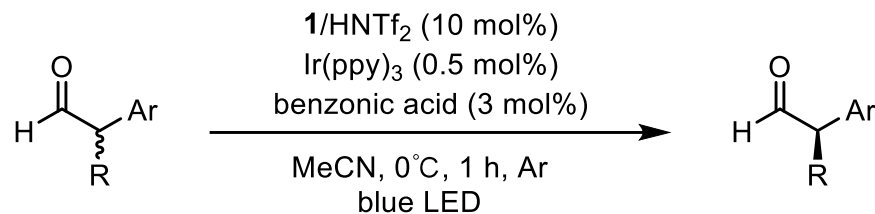
Deracemization at sp^3 -Hybridized Carbon



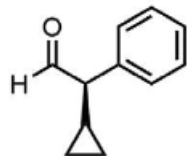
α -Deracemization of Ketones



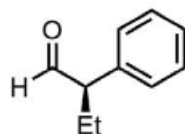
α -Deracemization of Ketones



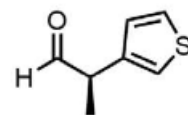
85%, 77% ee



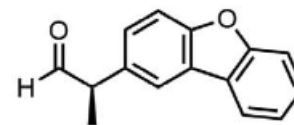
85%, 80% ee



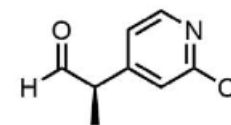
70%, 80% ee



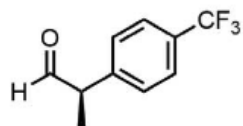
70%, 86% ee



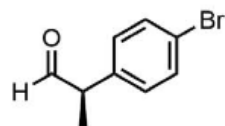
78%, 93% ee



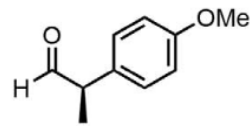
86%, 77% ee



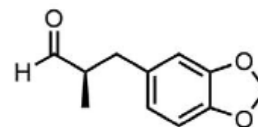
80%, 91% ee



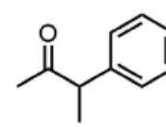
83%, 89% ee



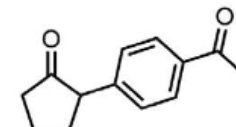
82%, 89% ee



73%, 5% ee



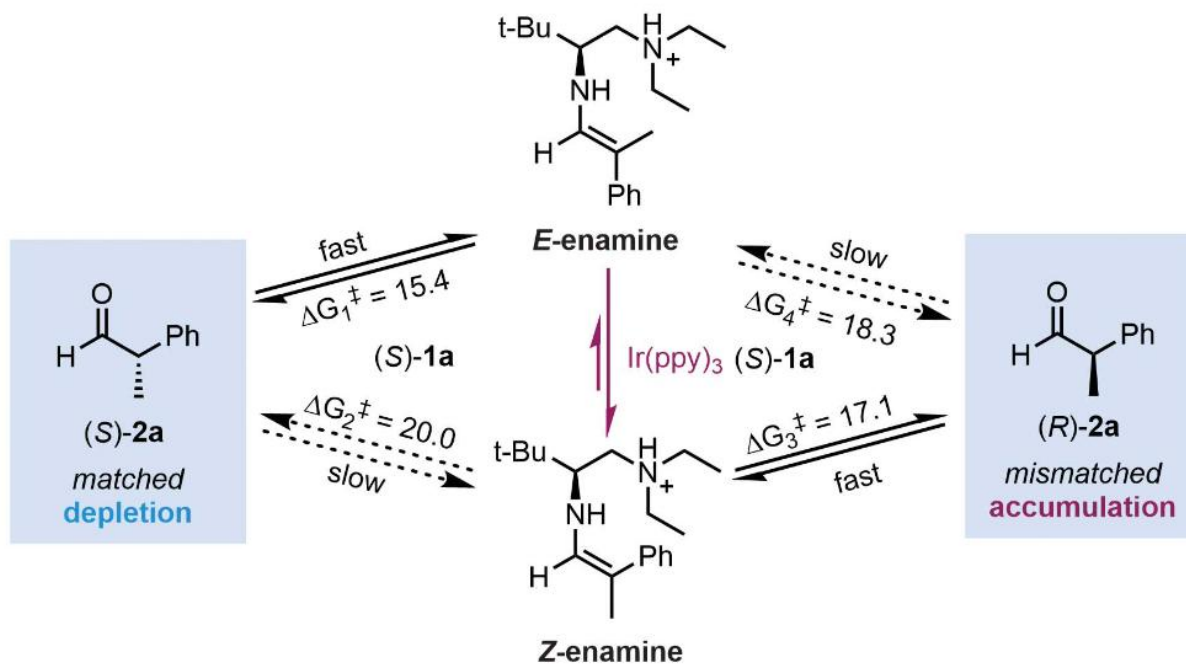
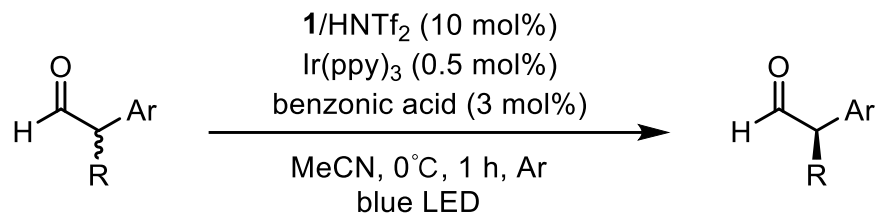
99%, rac



99%, rac

unsuccessful substrates

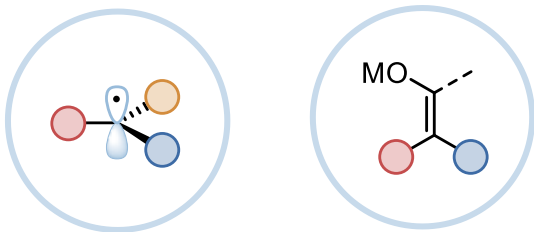
α -Deracemization of Ketones



Summary

■ Photocatalysis

■ Common intermediate



■ Energy input

■ Light energy

